This handbook applies to all students taking TR063 Physical Sciences. It provides a guide to what is expected of you on this programme, and the academic and personal support available to you. Please retain for future reference.

The information provided in this handbook is accurate at time of preparation. Any necessary revisions will be notified to students via email and the Science Course Office website (http://www.tcd.ie/Science). Please note that, in the event of any conflict or inconsistency between the General Regulations published in the University Calendar and information contained in course handbooks, the provisions of the General Regulations will prevail.

Produced by: The Science Course Office
Trinity College Dublin 2
Tel: +353 1 896 1970
Web Address: http://www.tcd.ie/Science/
Important information on COVID-19 restrictions and modes of teaching and learning

In order to offer taught programmes in line with government health and safety advice, teaching and learning in Semester 1 for your programme will follow a blended model that combines online and in-person elements to be attended on campus. This blended model will include offering online lectures for larger class groupings, as well as in-person classes for smaller groups: the differing modes of teaching and learning for particular modules are determined by your home School. Information on the modes of teaching and learning in Semester 2 will be available closer to the time.

Trinity will be as flexible as possible in facilitating late arrivals due to travel restrictions, visa delays, and other challenges arising from the COVID-19 pandemic. If you expect to arrive later than 28th September, please alert your course coordinator as early as possible.

For those students not currently in Ireland, according to current Government health and safety guidelines, please note that these students are expected to allow for a 14-day period of restricted movement after arrival and prior to commencement of their studies, and therefore should factor this into their travel plans.

For those students currently on the island of Ireland, we remind you of the Irish Government’s advice that all non-essential overseas travel should be avoided. If you do travel overseas, you are expected to restrict your movements for 14 days immediately from your return, during which time you will not be permitted to come to any Trinity campus. Therefore, as you are required to be available to attend College from the beginning of the new teaching year on 28 September, please ensure you do not return from travel overseas any later than 13 September.
Welcome to Physical Sciences

Welcome to Physical Sciences – your entry to Physics, Physics and Astrophysics and Nanoscience.

There are all sorts of reasons why you might have chosen Physical Sciences as your preferred degree course. You might have chosen out of a love of physics, or perhaps because you have a facility with mathematics. Perhaps you are driven by curiosity about the world in which we live, and the pleasure of figuring out how it works. In the Physical Sciences course you will have the opportunity to follow all these interests and more. We will help you cultivate a never-ending curiosity about the universe, whether this is at the largest scales of astrophysics, the smallest scales of particle physics, or the intermediate scales in which we live, and where our modern and future technology operates.

It may be that you have entered the course with the specific idea of graduating with one of the degrees, either in Physics, or in Physics and Astrophysics, or in Nanoscience. Perhaps you are planning to pursue a career in research in one of these areas. In Physical Sciences we would certainly encourage these dreams, but we will also prepare you for your future role in society and for many other careers, in industry and beyond. In every career move that you may make in the future your Physical Sciences degree will be useful, because it will have helped you develop transferable skills and attributes that are in demand by employers. Among these are problem solving skills, the ability to deal with complex mathematical and physical problems, and the use of the scientific method. You will learn scientific thought processes and critical thinking skills, and you will develop the Trinity Graduate Attributes that are important not just in many careers, but in many other areas of society.

Regardless of your personal motivations we would like to reassure you that you have chosen well, and that you will have many opportunities in the Physical Sciences course to reach your goals and potential – and that we will help you do this. This of course is with the proviso that you engage fully from the outset with the course, the materials, the laboratories, your peers and the academic and other staff you encounter. In that sense you must be prepared to work independently. Your degree in the Physical Sciences course begins today!

The first two years of the Physical Sciences course cover the most essential topics necessary for each of the three possible degree routes in the final two years. In these “Fresher” years you will study Physics and Mathematics and one other subject. The Physics course includes topics in astrophysics, statistics, mechanics, thermodynamics, electricity and magnetism, acoustics and optics, nuclear physics and quantum physics. The Fresher Mathematics includes topics in calculus, linear algebra, differential equations, and Fourier analysis. As part of your Physics modules you will spend three hours per week in experimental or computational laboratories learning coding skills through Python.
In the Sophister years all students will continue to develop foundational topics in physics to an advanced level, through courses on quantum mechanics, electromagnetism and statistical mechanics. This is applied to atomic physics, condensed matter physics and particle physics. In addition to this common core, students taking Physics can choose from a range of specialist courses, covering areas such as magnetism, nanoscience, semiconductors, modern optics, superconductivity, & astrophysics. Students specialising in Physics and Astrophysics instead take courses on stellar and galactic structures, planetary and space science, the interstellar medium, and on cosmology. Students specialising in Nanoscience will study thin films and polymers, spectroscopy of nanostructures, and further modules in solid state chemistry, materials chemistry as well as condensed matter physics.

In addition, students in the Physical Sciences stream will use computers for numerical modelling via the Python programming language and how to use a range of physical instruments for making measurements in the laboratory. Students in the third year take a course in communication skills which helps them to hone their oral and written presentation skills and shows them how to prepare a curriculum vitae. Third year students can meet graduates of the School of Physics through the ‘Wild Geese’ GradLink programme where they receive guidance on building their careers.

All students in the fourth year undertake a nine-week full time research project, which is carried out in a research laboratory in Trinity or in a research lab in another university or research institute. Students commonly travel to the US, UK, France, Germany and Australia for their final year projects. Students specialising in Physics and Astrophysics may travel to a telescope observatory.

Trinity Physical Science degrees are accredited by the Institute of Physics, the professional body for physicists in Ireland and the UK, as a basis for the professional status ‘Chartered Physicist’ (CPhys). Our graduates are always in high demand in Ireland, abroad and in modern high-tech industries, as well as in teaching. You may also find careers in academic institutions, government and industrial research organisations, production facilities or in the meteorological service. Diverse opportunities in electronics, telecommunications, biophysics, hospital and health physics, automation and computing are always available. The Physical Sciences courses will give you the opportunity to acquire problem solving skills which will be highly valued by your future employer. It could also give you useful primary training for a legal, managerial or actuarial career, where a technical background is attractive.

Professor Cormac McGuinness
Director, TR063 Physical Sciences Course
TR063 Physical Sciences overview and module selection.

Module choices will be made online. Prior to selecting modules, you should read this booklet, in particular pages 3-7 and then go to the TR063 Module Choice Form to select your modules. If you feel that you need assistance with your choices, please contact us at jfsco@tcd.ie and we will be happy to help.

Please note that choices you make in Junior Freshman year may influence your choices in the second semester of Junior Freshman year and Senior Freshman year.

Students must take 40 core credit modules (20 per semester) as follows:

<table>
<thead>
<tr>
<th>Module Code</th>
<th>Module Name</th>
<th>Credits</th>
</tr>
</thead>
<tbody>
<tr>
<td>PYU11P10</td>
<td>Physics 1</td>
<td>10</td>
</tr>
<tr>
<td>PYU11P20</td>
<td>Physics 2</td>
<td>10</td>
</tr>
<tr>
<td>MAU11S01</td>
<td>Mathematics for Scientists 1</td>
<td>10</td>
</tr>
<tr>
<td>MAU11S02</td>
<td>Mathematics for Scientists 2</td>
<td>10</td>
</tr>
</tbody>
</table>

Students will choose Open Modules to the value of 20 credits (10 per semester) from the following:

<table>
<thead>
<tr>
<th>Module Code</th>
<th>Module Name</th>
<th>Credits</th>
</tr>
</thead>
<tbody>
<tr>
<td>BYU11101</td>
<td>From Molecules to Cells</td>
<td>10</td>
</tr>
<tr>
<td>BYU11102</td>
<td>From Organisms to Ecosystems</td>
<td>10</td>
</tr>
<tr>
<td>CHU11101</td>
<td>General and Physical Chemistry</td>
<td>10</td>
</tr>
<tr>
<td>CHU11102</td>
<td>Introduction to Inorganic and Organic Chemistry</td>
<td>10</td>
</tr>
<tr>
<td>GSU11004</td>
<td>Spaceship Earth: Introduction to Earth System Science</td>
<td>10</td>
</tr>
<tr>
<td>GSU11005</td>
<td>Geology: A Beginner’s Guide to Planet Earth</td>
<td>10</td>
</tr>
</tbody>
</table>

Note that only certain patterns of choice of Open Modules are possible, and that the choice of Open modules in the Junior Freshman year then determines the Open Modules that must be taken in the Senior Freshman year.
Physical Sciences Moderatorships

In the Junior and Senior Freshman years TR063 students complete a course of study which will qualify them to compete for places in the following Moderatorships after the Senior Freshman year. The three available Moderatorship choices are:

- Physics
- Physics and Astrophysics
- Nanoscience

Moderatorship in Physics

In brief, a Moderatorship in Physics leads to a recognised Physics degree and encompasses the core physics subjects such as quantum mechanics, mechanics, thermodynamics, electromagnetism, oscillations and waves, condensed matter physics, atomic physics, relativity, nuclear structure, statistical physics, lasers and optics among others. The broad based Physics Moderatorship places an equal emphasis on diverse important subjects of nanoscience, magnetism, semiconductor device technology, photonics, nuclear and particle physics, superconductivity as well as aspects of astrophysics. It serves as the ideal launching point for a research or industrial career following any or all of these subject areas.

Moderatorship in Physics and Astrophysics

A Moderatorship in Physics and Astrophysics leads to a recognised Physics degree and encompasses the core physics subjects such as quantum mechanics, mechanics, thermodynamics, electromagnetism, oscillations and waves, condensed matter physics, atomic physics, relativity, nuclear structure, statistical physics, lasers and optics among others. The Physics and Astrophysics Moderatorship places astrophysics, stellar physics, stellar and galactic formation, cosmology and astronomical techniques first and foremost in this degree albeit at the expenses of subjects such as nanoscience, photonics, superconductivity and semiconductor device technology.

Moderatorship in Nanoscience

A Moderatorship in Nanoscience leads to a recognised Physics degree and encompasses the core physics subjects such as quantum mechanics, mechanics, thermodynamics, electromagnetism, oscillations and waves, condensed matter physics, atomic physics, relativity, nuclear structure, statistical physics, lasers and optics among others, as well as a recognisable core of physical, inorganic, organic and materials chemistry. The Nanoscience Moderatorship places the physics and chemistry of modern materials first and foremost which encompasses nanoscience, semiconductors and semiconductor device technology, photonics, materials chemistry, electrochemistry, polymers and photochemistry, all topics relevant to modern materials research, inclusive of energy materials, sensors and of microelectronics and any underpinning nanoscience.
Open Module Choices in Junior and Senior Freshman Years

The following table is indicative of the core modules and available Open modules in both Junior Freshman and Senior Freshman year of the TR063: Physical Sciences degree.

<table>
<thead>
<tr>
<th>Year 1: JUNIOR FRESHMAN</th>
<th>Year 2: SENIOR FRESHMAN</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>CORE MODULES – 40 credits 20/20</strong></td>
<td><strong>CORE MODULES – 40 credits 20/20</strong></td>
</tr>
<tr>
<td>Semester 1</td>
<td>Semester 2</td>
</tr>
<tr>
<td>PYU11P10 Physics 1</td>
<td>MAU11S02 Mathematics</td>
</tr>
<tr>
<td>PYU11P20 Physics 2</td>
<td>MAU11S03 Fourier Analysis</td>
</tr>
<tr>
<td>MAU11S01 Mathematics</td>
<td>MAU22S04 Vector Calculus</td>
</tr>
<tr>
<td><strong>OPEN MODULES – choose 20 credits 10/10</strong></td>
<td><strong>OPEN MODULES – choose 20 credits 10/10</strong></td>
</tr>
<tr>
<td>CHU11101 General and Physical Chemistry</td>
<td>CHU22101 Chemistry</td>
</tr>
<tr>
<td>CHU11102 Introduction to Inorganic and Organic Chemistry</td>
<td>CHU22102 Chemistry</td>
</tr>
<tr>
<td>GSU11004 Spaceship Earth: Introduction to Earth System Science</td>
<td>GSU22001 From Atoms to Rocks: Introduction to Geochemistry</td>
</tr>
<tr>
<td>BYU11101 From Molecules to Cells</td>
<td>BYU22201 From Molecules to Cells 2</td>
</tr>
<tr>
<td>BYU11102 Organisms to Ecosystems</td>
<td>BYU22202 From Cells to Organisms</td>
</tr>
<tr>
<td>MAU22S01 Multi-variable calculus</td>
<td>MAU22S03 Fourier Analysis</td>
</tr>
<tr>
<td>HEPoS History, Ethics, Philosophy of Science</td>
<td></td>
</tr>
</tbody>
</table>

Note that there are limitations on Open module choices across the Junior and Senior Freshman years due to prerequisites in the Senior Freshman year that depend upon or require Open modules in the Junior Freshman year. The possible patterns of Open Modules are shown in the diagram on the next page.
Moderatorships and Open Module Choice Diagram

Moderatorships in **Physics** or in **Physics and Astrophysics** are available to all students regardless of the choice of Open modules in the Junior Freshman and Senior Freshman years.

To qualify for the Moderatorship in **Nanoscience**, a student must take all available Chemistry Open modules in both semesters of the Junior and Senior Freshman years.

The following five patterns of Open Modules are available to students across the Junior Freshman and Senior Freshman years. These are denoted A, B, C, D and E and correspond to the indicated patterns on the TR063 Physical Sciences Junior Freshman module choice form.

<table>
<thead>
<tr>
<th>Moderatorship</th>
<th>Year 1: JUNIOR FRESHMAN</th>
<th>Year 2: SENIOR FRESHMAN</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Nano-science</strong></td>
<td>Semester 1</td>
<td>Semester 2</td>
</tr>
<tr>
<td>A</td>
<td>CHU11101 General and Physical Chemistry</td>
<td>CHU11102 Introduction to Inorganic and Organic Chemistry</td>
</tr>
<tr>
<td>D</td>
<td>BYU11101 From Molecules to Cells</td>
<td>BYU11102 Organisms to Ecosystems</td>
</tr>
</tbody>
</table>
Applications to choose a specific Moderatorship after the Senior Freshman year occur via a preferred Moderatorship choice form that will be available in Semester 2 of the Senior Freshman year. There are quotas and hence competition for in-demand Moderatorships.

All qualified Physical Sciences students will be able to proceed to a Moderatorship.

**TR063: PHYSICAL SCIENCES**

**CORE MODULES (mandatory) – 20 credits per semester**

| SEMESTER 1 – Michaelmas term  
| 5th October 2020 – 18th December 2020 | SEMESTER 2 – Hilary Term  
| 1st February 2021-23rd April 2021 |
|--------------------------------------|--------------------------------------|
| PYU11P10: Physics 1                | PYU11P20: Physics 2                |
| MAU11S01: Mathematics 1            | MAU11S02: Mathematics 2            |

**OPEN MODULES (optional): Students choose 10 credits from each semester**

<table>
<thead>
<tr>
<th>BYU11101: From Molecules to Cells</th>
<th>BYU11102: Organisms to Ecosystems</th>
</tr>
</thead>
<tbody>
<tr>
<td>CHU11101: General and Physical Chemistry</td>
<td>CHU11102: Introduction to Inorganic and Organic Chemistry</td>
</tr>
</tbody>
</table>

**Semester structure**

**Change of selected Open Modules**

If, after a couple of weeks, you feel that you have perhaps made the wrong choice of Open module, please seek advice immediately from your Tutor, Course Director or the Science Course Office. It may be possible for you to change from one module to another within Science, subject to permission from the Associate Dean of Undergraduate Science Education. If you do decide to change modules, then do so quickly - it can be difficult to try to catch up with work in a new module if you have missed more than two or three weeks of lectures. You should call into the Science Course Office if you wish to change modules.
TR063 Physical Sciences Core Modules

PYU11P10: Physics 1
Semester 1, 10 credits

The most fundamental foundational aspects of any physics education concern the motion of objects due to forces and how to mathematically describe these motions. Collective motions in response to forces lead to propagating physical waves, where similar mathematics can then describe electromagnetic waves or light. The first semester of your Physical Sciences education has an in-depth study of motion, forces, oscillation and light as the key physical concepts upon which to build. Of equal importance to the mathematical description of how the world we live in behaves, as described by a physical law, is an ability to make a measurement to verify or otherwise test the action of a physical law. Hence the physics laboratory plays a key role in the Physical Sciences education where the techniques of physical measurements are introduced together with the fundamentals of the experimental method and the manner in which the results of any experiment can be analysed.

Structure and contact hours
Lectures (4-5 hrs per week); practical laboratory (3hrs per week); online assignments (1 per week) and small group tutorials (1 hr every second week)

Lecture Topics
Introduction to Physics - 1 lectures (M. Möbius)
The Physics of Motion - 22 lectures (M Ferreira)
Waves and Optics I - 20 lectures (L Bradley)
Statistics - 10 lectures (M. Möbius).

Learning outcomes
On successful completion of this module students should be able to:

- Express in mathematical language the motion of a body under the action of forces.
- Describe wave motion and relate it to basic phenomena in light and sound.
- Understand sources of errors in measurements and calculate their propagation.
- Prepare a brief report, which includes an error analysis, of a simple physical experiment.

Syllabus

Introduction to Physics: 1 lecture
An introduction to the School of Physics and the JF Physics course.

The Physics of Motion: 22 lectures
Kinematics: velocity, acceleration, representation of motion through graphs, projectile motion, circular motion; Statics: forces, torque, equilibrium; Dynamics: Force-motion relations, Newton's laws, work, energy, linear and angular momenta, impulse, collisions, conservation laws
Waves and Optics I: 20 lectures

Statistics: 10 lectures
Systematic and random errors. Discrete and continuous distributions such as binomial, Poisson, Gaussian and Lorentzian. Moments of a distribution. Histograms and probability densities. Estimation of mean and standard deviation in a measurement. Error propagation and transformation of variables in probability distributions. Linear regression analysis, method of least squares, goodness of fit (Chi squared) and plotting techniques. Introduction to programming basics in Python

Methods of Teaching and Student Learning:
A mixture of lectures, small group tutorials, hands-on laboratory practicals and weekly on-line assignments based on both numerical and conceptual questions from the textbook are used in the delivery of this module. The lecture course follows the material in the textbook very closely with reading assignments clearly indicated to students as the lecture course progresses.

The practical sessions are structured to provide an introduction to the process of measurement, estimations of uncertainty (error) and propagation of errors as applied to physics experiments as well as introducing students to programming and data analysis through Python based computational physics experiments. Each experiment has its own specific learning outcomes and is structured to further clarify concepts met in the textbook and lectures thus reinforcing learning. Weekly homework assignments, typically alternating between topics, are submitted by students through an online system and corrected, with some limited feedback to the student available through the online system post deadline. The lecturer has oversight of the scores and responses to each assignment and can address these in subsequent lectures and tutorials.

Small group tutorials – in groups of 6-8 – meet with assigned academics every second week to introduce and practice the concepts of physics problem solving and the use of mathematics in physics and to develop physics insight in the students. These small group tutorials try to emphasise peer learning within the tutorial format and these problem solving activities provides an additional opportunity for the assigned academic to assess understanding and gauge the knowledge level of the students.

Finally, a number of lecturers use class based polling of student responses to questions using the available “clicker” technologies.

Methods of Assessment and Weighting
Examination – Written examination paper 60%; Laboratory Practical work 30%; online tutorial homework assignments 10%.
Reading List:
- University Physics - extended version with Modern Physics, by Hugh D. Young and Roger A. Freedman, Addison-Wesley, 2020, 15th end.
  Students do NOT buy this book - further information at first lecture of term.

Online Assignments:
Online assignments are submitted through the Mastering Physics system where electronic access is associated with the required/provided text book.
https://www.masteringphysics.com/site/login.html

Online Resources:
Software used in the practical laboratory – Logger Pro; as well as examples of Python code for analysis of data in the practical laboratory are available through the School of Physics website:
https://www.tcd.ie/Physics/study/current/undergraduate/Software-and-online-resources/

Module Website:
Visit http://www.tcd.ie/Physics/study/current/undergraduate for links to all Physics modules and to Blackboard for each module.
The motion and response of electrons due to electric and magnetic forces as well as the energies of electrons in atoms, molecules or metals determine almost all our interactions with our surroundings. The technological era is predicated on the motion of free electrons in electrical circuits, the coupling of motion to electric current and vice versa via magnetic interactions. The behaviour of electrons in atoms, molecules, metals and semiconductors is described by quantum theory which also describes electrons participating in the interaction of light and matter. An introduction to the quantum physics and quantum mechanics of light and electrons in atoms are the next foundational aspects of any physics education and are the heart of the second semester of your Physical Sciences education. Finally, our understanding and ability to observe the universe around us is through the interaction of light and matter, with the structure of the universe governed by the interaction of matter with matter. Gravitational and rotational dynamics determine the structure of the solar system and of the universe, and our knowledge of the universe is through the light we observe across all energy ranges. This is the last of the topics introduced in this first year of your education in the Physical Sciences. The physics laboratory continues in its key role in the Physical Sciences education with further training in experimental methods, analysis techniques and refining of your ability to describe the outcomes of an experiment.

Structure and contact hours:
Lectures (4-5 hrs per week); practical laboratory (3hrs per week); online assignments (1 per week) and small group tutorials (1 hr every second week)

Lecture Topics:
- Electricity and Magnetism - 20 lectures (J. Groh)
- Quantum Physics - 18 lectures (J. Pethica)
- Gravitation and Astrophysics - 12 lectures (A Vidotto)

Learning Outcomes:
On successful completion of this module students should be able to:
- Solve steady state time-varying electric current and electric potential problems
- Solve electrostatic problems using Gaussian Surfaces
- Describe how physics of matter and radiation is underpinned by quantum physics
- Develop the ideas of Newton's Law of Gravitation, and the motion of planets and satellites
- Describe the main properties of planets, exoplanets, the Sun and stars

Syllabus:

Electrostatics: electric charge, Coulomb's law, electric field, electric dipoles, Gauss's law, electric potential energy, voltage, electric polarization, capacitance, dielectrics, Electric current, resistance, Ohm’s law, electromotive force, power in electric circuits, Kirchoff's laws, RC circuits. Magnetism, magnetic field lines and flux; Lorentz force on moving charge; Energy of and torque on a current loop in a magnetic field; Biot-Savart Law illustrated by magnetic fields of a straight wire and circular loop; forces between current-carrying straight wires; Ampere’s Law in integral form.
Quantum Physics: 18 lectures

Gravitation and Astrophysics: 12 lectures

Methods of Assessment and Weighting:
Examination – Written examination paper 60%; Laboratory Practical work 30%; online tutorial homework assignments 10%.

Reading List:
- University Physics - extended version with Modern Physics, by Hugh D. Young and Roger A. Freedman, Addison-Wesley, 2020, 15th end.
  Students do NOT buy this book - further information at first lecture of term.

Online Assignments:
Online assignments are submitted through the Mastering Physics system where electronic access is associated with the required/provided text book.
[https://www.masteringphysics.com/site/login.html](https://www.masteringphysics.com/site/login.html)

Online Resources:
Software used in the practical laboratory – Logger Pro; as well as examples of Python code for analysis of data in the practical laboratory are available through the School of Physics website:
[https://www.tcd.ie/Physics/study/current/undergraduate/Software-and-online-resources/](https://www.tcd.ie/Physics/study/current/undergraduate/Software-and-online-resources/)

Module Website:
Visit [http://www.tcd.ie/Physics/study/current/undergraduate](http://www.tcd.ie/Physics/study/current/undergraduate) for links to all Physics modules and to Blackboard for each module.
Summary of Laboratory Practicals

Across Physics 1 and Physics 2 modules students complete 2 computational physics experiments (using Python) and 16 out of 20 available bench experiments for a total of 18 experiments performed by the student in the academic year. Many of the laboratory experiments are available on the bench in both semesters and thus the progress of students through the experiments differs from student to student with the exception of the computational physics experiments which all students complete. Students are required to record all data and information related to experiments in a hardback practical laboratory notebook which is assessed. A proportion of these experiments may be virtual experiments performed or data analysed at home in the academic year 2020/2021 but with virtual partners, possibly augmented by at-home experimental measurements. Some assessments may then also take place in a virtual interview, instead of at-bench interviews.

Laboratory Practicals

Introduction to Python
Python lab 1: Monte Carlo Approximation
Python lab 2: The Trajectory of a Projectile with Friction
Experiment 1: The Pendulum
Experiment 2: Energy Conservation
Experiment 3: Thin Lenses
Experiment 4: Density and the Principle of Archimedes
Experiment 5: Surface Tension
Experiment 6: Electrical Resistance
Experiment 7: DC Circuits
Experiment 8: Charging/Discharging a Capacitor
Experiment 9: Collisions and Momentum Transfer
Experiment 10: The Resonance Tube
Experiment 11: Leslie’s Cube
Experiment 12: Faraday’s Law
Experiment 13: Aperture and Depth of Field
Experiment 14: Interference and Diffraction
Experiment 15: The Geiger Counter
Experiment 16: Centripetal Acceleration
Experiment 17: The Photoelectric Effect
Experiment 18: The Bandgap of Germanium
Experiment 19: The Spectrometer
Experiment 20: AC circuits

Online Resources:
Software used in the practical laboratory – Logger Pro; as well as examples of Python code for analysis of data in the practical laboratory are available through the School of Physics website: https://www.tcd.ie/Physics/study/current/undergraduate/Software-and-online-resources/
Assessment of the laboratory

Half of a student’s experiments are assessed through an at-the-bench laboratory notebook assessment – the rest of the student’s experiments are assessed through written reports of the experiment. In all experiments both the laboratory notebook and the submitted experimental reports must include and require a complete data analysis, error estimation and statistical analysis and description and concise report of the outcomes of the experiment, and any inferences or conclusions that can be drawn from the outcome. A similar assessment requirement applies to the python based computational physics experiments, with the addition of assessment of the code used by the student. As a proportion of these experiments may be virtual experiments performed at home, some assessments may then also take place in a virtual interview, instead of at-bench interviews.

Junior Freshman Physics Coordinator
Professor Matthias Moebius
E-mail: mobiusm@tcd.ie
Ph: 01 896 1055

Executive Officer
Ms Helen O’Halloran
Ph: 01 896 4141
E-mail: physics@tcd.ie
MAU11S01: Mathematics for Scientists 1

Semester 1, 10 credits

Contact hours:
11 weeks, 6 lectures + 2 tutorials per week
This is the first semester module of a two semester sequence. It leads on to module MAU11S02 in the second semester.

Module Personnel

Prof. Kirk Soodhalter and Prof. Anthony Brown

Learning outcomes:

On successful completion of this module students will be able to

- Manipulate vectors to perform algebraic operations on them such as dot products and orthogonal projections and apply vector concepts to manipulate lines and planes in space $\mathbb{R}^3$ or in $\mathbb{R}^n$ with $n \geq 4$.
- Use Gaussian elimination techniques to solve systems of linear equations, find inverses of matrices and solve problems which can be reduced to such systems of linear equations.
- Manipulate matrices algebraically and use concepts related to matrices such as inevitability, symmetry, triangularity, nilpotence.
- Manipulate numbers in different bases and explain the usefulness of the ideas in computing.
- Use computer algebra and spreadsheets for elementary applications.
- Explain basic ideas relating to functions of a single variable and their graphs such as limits, continuity, inevitability, even/odd, differentiability and solve basic problems involving these concepts.
- Give basic properties and compute with a range of rational and standard transcendental functions, for instance to find derivatives, antiderivatives, critical points and to identify key features of their graphs.
- Use a range of basic techniques of integration to find definite and indefinite integrals.
- Apply techniques from calculus to a variety of applied problems.

Module content:

The content is divided in two sections, one for each lecturer.

Calculus with applications for Scientists

The lecturer for this part will be Prof. Anthony Brown. The main textbook will be [Anton] and the syllabus will be approximately 7 Chapters of [Anton] (numbered differently depending on the version and edition)

Chapter headings are

- Before Calculus (9th Ed) (was `Functions’ in the 8th edition);
- Limits and Continuity;
- The Derivative;
- The Derivative in Graphing and Applications;
- Integration;
- Exponential, Logarithmic and Inverse Trigonometric Functions;

Discrete Mathematics for Scientists

The lecturer for this part will be Prof. Kirk Soodhalter.

The order of the topics listed is not necessarily chronological. Some of the topics listed below linear algebra will be interspersed with linear algebra.
• **Linear algebra**  
  The syllabus for this part will be approximately chapters 1, 3 and parts of 10 from [AntonRorres].  
  - Vectors, geometric, norm, vector addition, dot product  
  - Systems of linear equations and Gauss-Jordan elimination;  
  - Matrices, inverses, diagonal, triangular, symmetric, trace;  
  - Selected application in different branches of science.

• **Computer algebra.**  
  An introduction to the application of computers to mathematical calculation. Exercises could include ideas from calculus (graphing, Newton’s method, numerical integration via trapezoidal rule and Simpsons rule) and linear algebra. We will make use of the computational software Mathematica which is used in many scientific applications.

• **Spreadsheets.** A brief overview of what spreadsheets do. Assignments based on Google docs.

• **Numbers.** An introduction to numbers and number systems e.g. binary, octal and hexadecimal numbers and algorithms for converting between them.

**Recommended reading list:**

[Anton]  
Or  
Single variable edition.

[AntonRorres]  

**Assessment details:**  
This module will be examined in a 3 hour examination in Trinity term. Assignments and tutorial work will count for 20% of the marks. There will be final examination in April/May counting for the remaining 80%. For supplementals, if required, the supplemental exam will count for 100%.

**Module Coordinators for MAU11S01**

- **Professor Kirk Soodhalter**  
  E-mail: ksoodha@maths.tcd.ie  
  ph. 01/896 8515

- **Professor Anthony Brown**  
  E-mail: anthony.brown@ucd.ie

**General enquires:**  
E-mail: mathdep@maths.tcd.ie
MAU11S02: Mathematics for Scientists 2

Semester 2, 10 credits

Contact hours:
11 weeks, 6 lectures + 2 tutorials per week

Module personnel
Prof. Miriam Logan and Prof Colm O'Dunlaing

Learning outcomes:

On successful completion of this module students will be able to:

- Apply definite integrals to various geometric problems.
- Apply various methods of integration.
- Use the concept of differential equations and methods of their solution.
- Use the concept of infinite series and their convergence; Taylor series.
- Use the concepts of parametric curves and polar coordinates.
- Define and calculate determinants by cofactor expansion and through upper triangular form.
- Use Cramer's Rule to solve linear equations.
- Use the Adjoint Matrix to invert matrices.
- Construct bases for row space, column space, and nullspace of a matrix.
- Construct orthonormal bases in three dimensions.
- Calculate the matrices of various linear maps.
- Compute linear and quadratic curves matching data through least squared error criterion.
- Calculate eigenvalues and eigenvectors for 2x2 matrices, with applications to differential equations.
- Derive probability distributions in simple cases.
- Apply the Binomial Distribution.
- Compute the conditional probability $P(A_i | D)$ given $P(D|A_i)$.
- Apply the Poisson distribution to traffic-light queuing problems.
- Apply continuous distributions, Normal, chi-squared, Student's t-distribution.
- Obtain confidence intervals for mean and standard deviation.
- Apply the Central Limit Theorem to approximate the binomial distribution for large $n$.
- Perform basic hypothesis testing.
Module content:

The content is divided in two sections, one for each lecturer.

Calculus with applications for Scientists

The lecturer for this part will be Prof Miriam Logan

- Application of definite integrals in geometry (area between curves, volume of a solid, length of a plane curve, area of a surface of revolution).
- Methods of integration (integration by parts, trigonometric substitutions, numerical integration, improper integrals).
- Differential equations (separable DE, first order linear DE, Euler method).
- Infinite series (convergence of sequences, sums of infinite series, convergence tests, absolute convergence, Taylor series).
- Parametric curves and polar coordinates.

Discrete Mathematics for Scientists

The lecturer for this part will be Prof Colm O Dunlaing

Module Content:
Linear Algebra

- This reference for this part of the course will be (AntonRorres). The syllabus will be approximately chapters 2, 5, section 4.2 and a selection of application topics from chapter 11 of (AntonRorres).
- Determinants, Evaluation by Row Operations and Laplace Expansion, Properties, Vector Cross Products, Eigenvalues and Eigenvectors;
- Introduction to Vector Spaces and Linear Transformations. Least Squares Fit via Linear Algebra;
- Differential Equations, System of First Order Linear Equations;
- Selected Application in Different Branches of Science;

Probability

- Basic Concepts of Probability; Sample Means; Expectation and Standard Deviation for Discrete Random Variables; Continuous Random Variables; Examples of Common Probability Distributions (binomial, Poisson, normal) (sections 24.1 - 24.3, 24.5 - 24.8 of (Kreyszig).

Recommended reading lists:

(Stewart)

(Anton)
- Combined edition:
Module Prerequisite:
MAU1S001 Mathematics for Scientists 1 (First Semester)

Assessment details:
This module will be examined in a 3 hour examination in Trinity term. Continuous assessment in the form of weekly tutorial work will contribute 20% to the final grade at the annual examinations, with the examination counting for the remaining 80%. For supplementals if required, the supplemental exam will count for 100%.

Module Coordinators for MAU1S02

Professor Miriam Logan
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Professor Colm O’Dunlaing
Ph: 01 896 1948
E-mail: odunlain@maths.tcd.ie

General enquires: E-mail: mathdep@maths.tcd.ie
Module learning aim:
This module aims to provide an introduction to molecular and cellular biology, and will therefore include key topics in Cell Biology, Biochemistry, Genetics, and Microbiology. A description of the possible origin of life, from the abiotic world to single-celled and multicellular organisms will be given, and the ultrastructure of the prokaryotic and eukaryotic cells will be covered in detail. The properties and functions of the major classes of biochemicals found in living systems (carbohydrates, lipids, proteins and nucleic acids) will be described, the structure and function of membranes and organelles, and the chemical basis of metabolism and energy transfer in the cell. Students will then be introduced to basic concepts in Genetics, how the information contained in DNA (genes) is expressed, replicated and inherited. Finally, the sheer diversity of life forms, from viruses to prokaryotic and eukaryotic microorganisms, to more complex plant and animal life forms will be described. Students will also study cell and virus structure, cell growth and viral replication, agents of infectious diseases, and host immunity.

Learning outcomes:
On successful completion of the module, students will be able to:
1. Provide an account of the cellular basis of life: from its origins in the abiotic world, to the evolution of unicellular and multicellular organisms.
2. Describe the diversity of life forms: including viruses, prokaryotes (bacteria), archaea, and eukaryotes (unicellular organisms, animals and plants).
3. Provide an account of the chemical basis of life and the biochemistry on which living systems depend: the properties and functions of the major classes of biomolecules, the structure and function of membranes and organelles, and the chemical basis of metabolism and energy transfer.
4. Describe how the information contained in DNA (genes) directs the construction and growth of an organism, and how this information is replicated and transmitted from one generation to the next (inheritance; genetics).
5. Employ a range of laboratory techniques, demonstrating the development of practical scientific skills, knowledge of experimental design and the interpretation of results.
6. Apply the scientific method as a fundamental approach to experiment-based investigations, critical analysis of data, and problem solving.
Module content:

Lectures

**Lecture 1: Module: Introduction, objectives and overview.**
- Module Coordinator: (Professor Tony Kavanagh)

**Section 1: Origin of Life - Cellular basis of life - Diversity of Life Forms:**

**Lecture 2 and 3: Origin of Life** (Professor Luke O’Neill)
- What is Life? How did it arise?
- The Origin of Life from a chemical and cellular perspective. The abiotic world
- The prebiotic world. Miller-Urey experiment. The first cell
- Photosynthesis and oxygen – mass extinction. Origin of first eukaryotic cell
- Multicellular life. Cell specialization

**Lecture 4 – 6: Cellular basis of life** (Professor Fred Sheedy)
- Cell structure – prokaryotes, archaea, eukaryotes - animal and plant
- Organelles & their prokaryotic origin – mitochondria, chloroplasts
- mitosis and meiosis – cell division – regulation of cell division (introduction)

**Lecture 7 – 12: Diversity of Microbial Life** (Professor Alastair Fleming)

**Lecture 7 - The Tree of Life** (Professor Alastair Fleming)

**Lecture 8 - Bacteria** (Professor Alastair Fleming)
- Cell structure, morphology, function and habitat.

**Lecture 9- Fungi & Protists** (Professor Alastair Fleming)
- Fungal cell structure, morphology, function and habitat.
- Protist cell structure, morphology, habitat and life-cycles: extreme cell diversity

**Lecture 10: The Archaea** (Professor Alastair Fleming)
- A third way of life; features of both prokaryotes and eukaryotes: cell structure, morphology and function. Extremophiles

**Lecture 11: Viruses** (Professor Alastair Fleming)
- Alive? Structure, function and habitat

**Lecture 12 – Relationship between life forms:** (Professor Alastair Fleming)
- The good, the bad and the ugly. Concepts of symbiosis and parasites. Plant and animal diseases

**ONLINE ASSESSMENT via MCQ**
**Section 2: Chemistry of Life:**

**Lecture 13 - Introduction to biochemistry:**

**Structural principles for small molecules** (Professor Luke O’Neill)
- Elements and chemical groups commonly found in nature
- Bonds, bond energies, bond lengths in nature
- Asymmetry: right and left-handed molecules etc.
- Typical forces between molecules and chemical groups in nature
- Four basic classes of Biomolecules: amino acids, nucleotides, carbohydrates & lipids

**Lecture 14: Nucleotides, Amino acids and peptides** (Professor Luke O’Neill)
- Classes of nucleic acids (DNA, RNA), Chromatin and chromosome structure, Properties of amino acids: chemical features and physical properties of the R-groups
- The peptide unit and peptide bond
- Introduction to polypeptides & concept of folding

**Lecture 15: Proteins and protein structure** (Professor Ken Mok)
- Concept that shape dictates function
- Hierarchical organization of protein structure: concept of primary, secondary, tertiary and quaternary structure.
- Introduction to forces that stabilize protein structure
Lecture 16: Protein function (Professor Ken Mok)
- Major functional classes of protein
- Introduction to bioinformatics: Proteins and evolution; relationships between proteins: similarity and identity.

Lecture 17 and 18: Enzymes: the catalysts of life (Professor Vincent Kelly)
- Enzyme structure & function
- Enzyme reaction mechanism (co-factors and vitamins)
- Enzyme kinetics
- Regulation of enzyme activity

Lecture 19: Lipids and membranes (Professor Vincent Kelly)
- Lipid structures: fatty acids, phospholipids etc
- Membranes: chemical and physical properties
- Membrane proteins
- Transport across membranes
- Concept of compartmentation and membrane traffic

Lecture 20 and 21: Metabolism & major metabolic pathways (Professor Vincent Kelly)
- The starting point: introduction to carbohydrates and fatty acids
- Organization, energetic principles, key steps and links between the main metabolic pathways.
- Glycolysis, TCA cycle, beta oxidation
- Outline of the reversing catabolic pathways: gluconeogenesis and fatty acids synthesis.

Lecture 22: Mitochondria & Respiration (Professor Vincent Kelly)
- Mitochondria, redox reactions and energy transduction
- Electron transport and the electron transport chain
- Oxidative phosphorylation
- Coupling of oxidation to phosphorylation: chemiosmotic view of energy transduction (in brief).

Lecture 23: Chloroplasts and Photosynthesis (Professor Vincent Kelly)
- Chloroplasts: architecture and function
- Overview of the light and dark reactions of photosynthesis

ONLINE ASSESSMENT via MCQ
Section 3: Biological Information: Genetics, Heredity and DNA

Lecture 24: Introduction to Genetics (Professor Jane Farrar)
- Introduction to the course content: An outline of some core concepts from classical genetics to the present will be presented. A whistle stop tour of key discoveries in the history of genetics.

Lecture 25: Mendelian Genetics (Professor Jane Farrar)
- Mendel’s laws: the 1st law of segregation and the 2nd law of independent assortment using monohybrid and dihybrid crosses; concepts relating to genetic analysis and the use of model systems will be described. Inheritance patterns for single gene disorders will be presented - pedigree analysis.

Lecture 26 and 27: Linkage and recombination (Professor Jane Farrar)
- Meiosis and the role of ‘crossing over’ in gene mapping (meiosis covered in detail elsewhere). A brief recap regarding Mendelian genetics – for example, highlighting that genetic linkage breaks Mendel’s 2nd law of independent assortment (refer back to L27)
- Outline of key concepts underlying the generation of genetic maps. Reference will be made to some classical work by Sturtevant / Morgan.

Lecture 28: Identification of DNA as hereditary material (Professor Jane Farrar)
- Key experiments establishing DNA as the genetic material; bacterial transformation and its significance (Griffith / Avery, McLeod & McCarthy / Hershey-Chase); the concept of horizontal gene transfer (mechanisms transformation, conjugation, transduction). Differences in vertical and horizontal gene transfer.

Lecture 29: Quantitative Genetics (Professor Jane Farrar)
- An overview of some concepts relating to discrete variation versus continuous variation. Experiments demonstrating that quantitative traits are inherited.
- Examples of some quantitative traits in humans. Concepts regarding the use of GWAS to elucidate the genetics architecture of complex traits using an example of one or more disorders.

Lecture 30: DNA, Structure and Function (Professor Tony Kavanagh)
- The double helix - discovery of the structure of DNA – DNA composition - DNA replication – semi-conservative replication, replication forks, leading and lagging strand synthesis, DNA polymerases
- DNA replication in prokaryotes and eukaryotes.

Lecture 31 -33: Information flow in the cell - The Central Dogma (Professor Tony Kavanagh).
- Transcription, RNA Polymerases in prokaryotes and eukaryotes.
- Promoters, repressors, terminators – the lac operon; transcription factors, enhancers.
- Decoding the information in mRNA:
- Translation (Protein synthesis).
- Ribosomes in prokaryotes and eukaryotes, tRNAs and aminoacyl tRNA synthetases, the genetic code; translation;
- Introduction to the regulation of gene expression – positive and negative regulation

Lecture 34: DNA – Mutation and its consequences (Professor Tony Kavanagh)
- Mechanisms by which mutations are generated - including errors in DNA replication; the action of chemical and physical mutagens; and errors in chromosome construction and distribution.
- An outline of the different types of mutation (missense, nonsense, frameshift mutations) and their molecular consequences in relation to gene expression and protein function. Mutations causing inherited diseases and cancer.
- DNA repair – mechanisms of DNA repair; repair deficiency and disease.

ONLINE ASSESSMENT via MCQ
Recommended reading lists:

(1) Campbell Biology, 11th Edition By Lisa A. Urry, Michael L. Cain, Steven A. Wasserman, Peter V. Minorsky, Jane B. Reece (Published by Pearson (2019))

Contact Hours/Methods of Teaching and Student Learning.

65 hours consisting of a mixture of lectures, tutorials and hands-on laboratory practicals. The practical sessions are formatted in order to further clarify concepts thus reinforcing learning. A tutorial in essay writing will help students prepare for the examination.

Method of assessment

Continuous assessment 50%

- Laboratory practical assessment: Four components: engagement = 5% of module, Assignment 1 (open book) covering practicals 1 to 5 inclusive = 5% of module, Assignment 2 (open book) covering Practicals 6 to 8 inclusive = 5% of module, end of module online assessment of all practicals (closed book) = 20% of module.
- There will be three online open book assessments of lecture material each worth 5%

Written Examination 50%

One written examination paper of 1.5 hour duration.

There will be three sections on the examination paper:

- Section 1 will have two questions on Section 1 (Origin of Life) you are required to answer one question from this section
- Section 2 will have two questions on Section 2 (Chemistry of Life) you are required to answer one question from this section
- Section 3 will have two questions on Section 3 (Biological Information) you are required to answer one question from this section
- All questions carry equal marks

Contacts

Module Coordinator: Molecules to Cells 1
Professor Tony Kavanagh  
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Course Director TR060: Biological and Biomedical Sciences:
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Freshman Biology Teaching Coordinator  
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Executive Officer  
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Phone: 01 896 1117
Module learning aim:
Organisms to Ecosystems I aims to introduce students to the biology of individuals, species, populations and ecosystems, and explore how humans interact with other living organisms. It will cover the developmental biology of organisms, their physiology, brain function and the evolutionary and ecological responses of organisms to their environment. Topics incorporate the diversity of life and its biological development, interactions between organisms and their environment, the biological context of climate change, human impacts on the environment, future food sustainability, urban ecology, ecosystem services and the value and conservation of biodiversity. Topics are arranged in three sections: 1) Multicellularity and Development, Physiology, Behaviour and Neuroscience, 2) Evolution: Adaptation, Populations and Biodiversity, and 3) Ecology and Environment.

A mixture of lectures, tutorials and hands-on laboratory practicals are used in the delivery of this module. There will be one - ecology practical - on a field site outside of campus.

Learning outcomes
On successful completion of this course, the student will be able to:
1. Outline the major steps involved in how complex animal and plants are formed and be able to relate the morphological changes that occur to the molecular and cellular changes that underlie and drive embryo and organ development
2. Describe the concept of homeostasis at the cell, organ and organism level; give examples of the functional interrelationships that exist between cells, organs and systems; provide an account of how organisms can sense change in the environment
3. Describe the basic principles by which the brain functions and outline key experimental steps and informative clinical cases that have elucidated our current understanding of brain function.
4. Recognise the diversity of life on earth and describe how it evolved over geological time scales
5. Describe the ecological relationships between individuals, populations, communities and ecosystems, and between organisms and their environment
6. Recognise how humans can positively and negatively influence other living organisms and their environment and understand the value of other living organisms for humans
7. Demonstrate practical, numerical and analytical skills
8. Collate, synthesise, organise and present information in written reports
## Module content:

**Lectures**

**Lecture 1**

**Lecture 1** - Module: Introduction, objectives and overview.  
Module Coordinator Professor Trevor Hodkinson  
**Section 1A - Multicellularity and Development (Professor Rebecca Rolfe)**

<table>
<thead>
<tr>
<th>Lectures</th>
<th>Topic and Content</th>
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</table>
| Lecture 1 | **Lecture 1** - Module: Introduction, objectives and overview.  
Module Coordinator Professor Trevor Hodkinson  
**Section 1A - Multicellularity and Development (Professor Rebecca Rolfe)** |

### 5 lectures

| Lecture 2 | Introduction to development  
- core concepts / model organisms / analysis of development (morphology, genetic, biochemical) |
| Lecture 3 | Embryogenesis and morphogenesis  
- germ layers |
| Lecture 4 | Intercellular communication, determination, potency, axis formation  
- anterior-posterior, dorsal-ventral. |
| Lecture 5 | Pattern formation  
- morphogens, gradients and thresholds. |
| Lecture 6 | Differential gene expression  
- temporal and spatial, master regulators |

### 6 lectures

| Lecture 7 | Form and Function  
- Functional characteristics of living things  
- Specialisation of cells/tissues/organs to fulfill specific functions |
| Lecture 8 | Homeostasis.  
- The concept of the internal environment. Composition, temperature, pH etc. of body fluids. Maintenance of homeostasis by cooperation of different physiological systems. Feedback and feed-forward |
| Lecture 9 | Physiological regulation of function.  
- Fundamentals of nervous and endocrine control of function and comparison of speed and modes of action. How an individual organism senses and responds to changes in the external and internal environments. |

**Section 1B: Physiology (Professor Áine Kelly)**

| Lecture 7 | Form and Function  
- Functional characteristics of living things  
- Specialisation of cells/tissues/organs to fulfill specific functions |
| Lecture 8 | Homeostasis.  
- The concept of the internal environment. Composition, temperature, pH etc. of body fluids. Maintenance of homeostasis by cooperation of different physiological systems. Feedback and feed-forward |
| Lecture 9 | Physiological regulation of function.  
- Fundamentals of nervous and endocrine control of function and comparison of speed and modes of action. How an individual organism senses and responds to changes in the external and internal environments. |

**Behaviour and Neuroscience (Professor Tomas Ryan)**

| Lecture 10 | Pre-neuroscience history of mind/brain ideas  
- Cartesian dualism and materialist and non-materialist explanations of mind. The brain as the substrate of mind. The effects of head trauma on behaviour and memory, anatomy of the human/mammalian brain, functions in behaviour and in homeostasis, overview of human brain regions and attribution of various regions to broad functions (evidence from lesions, imaging). |
| Lecture 11 | Fundamentals of nervous system structure and function  
- Reticular vs. neuron theory, nervous system as electrically active, Helmholtz and excitable neurons, action potentials & synaptic transmission. |
| Lecture 12 | Introduction to the biology of memory storage  
- Challenges of integrating neurobiology and brain function at multiple levels. Reductionism and correlation vs. causation. The biology of memory storage. |
| Lecture 13 | Summary of key concepts: Q&A Profs. Rolfe, Kelly and Ryan |

**ONLINE ASSESSMENT via MCQ**

**Section 2: Evolution: adaptation, populations, biodiversity (Professor Trevor Hodkinson)**

| Lecture 14 | Short history of life  
- timeline, major groups, diversity. |
Lecture 15 - Selection/modern synthesis
- adaptation

Lecture 16 - Species
- Definitions, taxonomy, diversity, species rich groups

Lecture 17 - Speciation
- allopatric, sympatric, adaptation, radiations, key innovations

Lecture 18 – Extinction
- fossils, global change (climate, atmosphere, tectonic)

Lecture 19 - Phylogeny, homology, convergence, reversals, methods

Lecture 20 - Genetic basis of selection (Professor Aoife McLysaght)

Lecture 21 - Genetic basis of evolution
- Molecular variation, neutral theory, drift

Lecture 22 - Genetic basis of evolution
- Molecular evolution of population genetic variation

Lecture 23 - Genetic basis of evolution
- Population genetic variations

Lecture 24 - Human evolution

Lecture 25 - Summary of key concepts: Q&A Profs. Hodkinson & McLysaght

ONLINE ASSESSMENT via MCQ

Section 3: Ecology and Environment (Professor Jennifer McElwain & Professor Yvonne Buckley)

Global ecology (Professor Jennifer McElwain)

Lecture 26 - Global ecology and climate change
- Future climate change – global challenges – projections
- Pest diseases, human physiology, how to predict
- Need to understand fundamentals of ecology to address these global challenges

Lecture 27 - Biomes, niches
- Introduction to biomes, what shapes biome distribution? climate change
- concepts – climate niches / fundamental versus realized niche
- Challenge of predicting future ecological responses to climate change

Populations to communities: challenges & solutions (Professor Yvonne Buckley)

Lecture 28 - Commonness, rarity and population processes (Professor Yvonne Buckley)
- Extinction or persistence are processes that operate at the population level.
- Introduction to concepts of abundance and rarity, competition, dispersal, demography and its application to conservation (incl. endemism and invasions)

Lecture 29 - Conservation
- applications of population biology at the species level, including prioritizing species for conservation management, assessing threat and red listing

Lecture 30 - Trophic cascades and rewilding
- What is a community, energy flow, applications of community ecology to conservation and rewilding challenges. Consumption, facilitation & predation

Lecture 31 - Constructing ecosystems and conservation
- In the Anthropocene humans have constructed new ecosystems, what are they, where do we find them and what are their values? Contrast with “natural” ecosystems

Lecture 32 - Urban ecology
- how have organisms adapted to living in urban environments? How can we better design our cities and buildings to gain more value from nature and support biodiversity?
Lecture 33 - Ecosystem services and natural capital
- Nature provides many valuable ecosystem services supported by natural capital
- Introduction to the concepts and controversies surrounding the ecosystem services and natural capital concepts

Food and feedbacks (Professor Jennifer McElwain)
Lecture 34 - Food: environmental impacts and ecological process
- Food security- ecological concepts- productivity- energy flows through ecological systems/basic concepts of biogeochemical cycles.

Lecture 35 - Future food and a changing planet
- Food security
- Ecological concepts, human population increase, projections for future productivity.

Lecture 36 - Biosphere feedbacks on climate system
- Introduction to biological feedbacks on the climate system: at global level
- Carbon sequestration/ transpiration/ water budget, within biomes
- Fire feedbacks/rain seeding. Nature based solutions to climate mitigation and adaptation
- Green and blue solutions – cities etc., ‘The Martian’ closed systems

Lecture 37 - Summary of key concepts: Q&A Profs. Buckley & McElwain
Online Assessment via MCQ

Lecture 38 - Module: Overview and integration of all concepts
Module Coordinator - Trevor Hodkinson

Recommended reading lists:
1. Campbell Biology, 11th Edition By Lisa A. Urry, Michael L. Cain, Steven A. Wasserman, Peter V. Minorsky, Jane B. Reece (Published by Pearson (2019))

Contact Hours/Methods of Teaching and Student learning.
65 hours consisting of a mixture of lectures, tutorials and hands-on laboratory practicals. The practical sessions are formatted in order to further clarify concepts thus reinforcing learning.

Method of assessment
Continuous assessment 50%
- Laboratory practical assessment: 35%
- Online Assessment via multiple choice questions (MCQ) 15%: there will be three MCQ online assessments, each worth 5%

Written Examination 50%
One written examination paper of 1.5 hour duration
There will be three sections on the examination paper:
- Section 1 will have 2 questions on Section 1 (Multicellularity and Development; and Physiology) students are required to answer 1 question from this section
- Section 2 will have 2 questions on Section 2 (Evolution: adaptation, populations, biodiversity) students are required to answer 1 question from this section
- Section 3 will have 2 questions on Section 3 (Ecology and Environment) students are required to answer 1 question from this section
The three sections are equally weighted.
Contacts:

**Module Coordinator: From Organisms to Ecosystems**

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**Course Director TR060: Biological and Biomedical Sciences:**

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Freshman Biology Teaching Coordinator  
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Executive Officer  
E-mail: BTC.Administrator@tcd.ie  
Phone: 01 896 1117
CHU11101: General and Physical Chemistry

Semester 1, 10 credits

**Rationale and Aims**: To provide a general introduction to chemistry and physical chemistry and equips the student with the knowledge to understand the basic concepts in chemistry, understanding of the building principles of matter, chemical bonding and molecular structure, an introduction to thermodynamics, electrochemistry, acid/base reactions and to the chemistry of liquids, solids and solutions.

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<td>• Motivation for studying chemistry; physical states of chemical matter; classification of matter, physical and chemical properties of pure substances and mixtures; extensive and intensive properties; chemical analysis.</td>
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<td></td>
<td>• Measurements and units; the international system of units; derived units, the reliability of measurements and calculations; significant figures in simple calculations.</td>
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<tr>
<td></td>
<td>• Structure and building principles of atoms; element symbols; masses and the mole; introduction to the periodic table; brief introduction into the structure of the electron shell; ionisation energy and electron affinity.</td>
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<td>• Law of conversation of mass; law of definite composition; bonding in chemical substances; ionic bonding; covalent bonding; weak bonding; molecules and solid state structures; electronegativity; the periodic table.</td>
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<td></td>
<td>• Chemical nomenclature of inorganic compounds; stoichiometry; mole, molarity and concentration; interpreting stoichiometric coefficients; example calculations.</td>
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<td>• Chemical reactions; symbolizing reactions; balancing equations; limiting reagents and yields; role of water in chemical reactions; important classes of chemical reactions; precipitation reactions; examples of precipitation reactions in chemistry net ionic equations.</td>
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<tr>
<td></td>
<td>• Introduction into acid and base reactions; acid-base titration, introduction into oxidation and reduction reactions; oxidation number and electron transfer; oxidizing and reducing agents; half-reactions.</td>
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<tr>
<td></td>
<td>• Oxidation and reduction reactions, electron transfer; oxidizing and reducing agents; half-equations.</td>
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<td>The Electronic Theory of Chemistry</td>
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<td>• Wave properties of particles.</td>
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<td>• The structures of many-electron atoms. Orbital energies.</td>
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<td>• The building-up principle.</td>
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<td>• A survey of periodic properties.</td>
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<td>• Assessing the charge distribution.</td>
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<td>• Polarization. Ionic and atomic radii.</td>
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<td>• Variable valence.</td>
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<td>• Covalent bonds.</td>
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<tr>
<td>• The electron-pair bond. Lewis acids and bases.</td>
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<tr>
<td>• The Shapes of Molecules.</td>
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<tr>
<td>• Valence Shell Electron Repulsion theory.</td>
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<tr>
<td>• The arrangement of electron pairs.</td>
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<td>• Polar molecules.</td>
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<tr>
<td>• The orbital model of bonding.</td>
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<td>• Hybridization.</td>
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<td>• Molecular orbitals.</td>
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<tr>
<td>• Bonding in Period 2 diatomic molecules.</td>
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<tr>
<td>• A perspective on chemical bonding.</td>
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<table>
<thead>
<tr>
<th>5-12 (24 L) Introduction to Physical Chemistry</th>
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<tbody>
<tr>
<td>• The ideal gas law</td>
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<tr>
<td>• Kinetic molecular theory of ideal gases</td>
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<td>• Differences between real and ideal gases</td>
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<td>• The First Law of Thermodynamics</td>
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<tr>
<td>• Internal Energy, Enthalpy and Calorimetry</td>
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<td>• Cp and Cv, expansion/compression of gases. Adiabatics.</td>
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<td>• The Second Law of Thermodynamics: entropy</td>
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<td>• The Carnot cycle</td>
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<td>• Gibbs’ Free Energy</td>
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<td>• Chemical Equilibrium</td>
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<td>• Boltzmann’s Factor</td>
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<td>• Acids-Bases and Titrations</td>
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</tbody>
</table>
Reading list/ Indicative Resources

- Chemistry & Chemical Reactivity Hardcover by Paul Treichel, John Kotz, John Townsend, David Treichel; Publisher: Brooks Cole; 9 ed.

Methods of Teaching and Student Learning
A mixture of lectures, tutorials and hands-on laboratory practicals are used in the delivery of this module. The practical sessions are formatted in order to further clarify concepts thus reinforcing learning. A weekly hour of tutorial problem solving activities provides an additional opportunity for the lecturer to assess understanding and gauge the knowledge level of the students.
All lecture notes and problem sheets and a selection of self-assessment quizzes are available for students on Blackboard.

Learning outcomes
On completion of this module the student should be able to:

- Explain, using appropriate terminology and physical units, basic concepts in chemistry, including precipitation and redox reactions.
- Analyse bonding and atomic molecular structure
- Apply the ideal gas law to calculations of gas properties
- Describe the principles underpinning the kinetic theory of gases
- Analyse and identify the main types of intermolecular forces
- Identify and explain the principal features of the phase diagrams of pure compounds, including pressure dependence of melting and boiling points, triple point and critical point, and variation of vapour pressure with temperature.
- Calculate chemical equilibria and illustrate the key concepts, including variation of components with concentration, temperature and pressure
- Discuss simple acid/base chemistry and apply to solution equilibria
- Illustrate the basic concepts of an electrochemical cell, including half-cell reactions, cell potential and reaction free energy and be able to determine these properties as well as concentration dependence
- Describe the main classes of the solid-state structure; cubic- and hexagonal close packing; body-centred and face-centred cubic structures. Octahedral and tetrahedral holes, coordination numbers, the Born-Haber cycle, lattice energy
- Identify, describe and analyse the factors affecting solubility.
• Define and explain colligative properties, including Raoult’s Law and the calculation of molecular weights
• Understand and apply the concepts underlying the First and Second Laws of Thermodynamics to numerical problems

Assessment details:
This module will be examined via a combination of in-course assessments (40% of the final mark) and a 3 h paper at the end of semester 1 (60% of the final mark).
CHU11102: Introduction to Inorganic and Organic Chemistry
Semester 2, 10 credits

## Content Layout

<table>
<thead>
<tr>
<th>Teaching Week</th>
<th>Topic</th>
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<tbody>
<tr>
<td>1-8 (28 L)</td>
<td>Introduction to Organic Chemistry</td>
</tr>
<tr>
<td></td>
<td>- Alkanes, isomers, homologous series, IUPAC nomenclature, physical properties and molecular size, the tetrahedral carbon atom, shapes of organic molecules, alicyclic rings, concept of bond strain, conformations of ethane and of the cyclohexane ring, chair and boat forms and their relative stabilities, axial and equatorial bonds.</td>
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<tr>
<td></td>
<td>- Alkenes, nomenclature, the double bond as an electron rich centre mechanism of electrophilic addition of hydrogen halides, water, and halogens to the double bond, Markownikoff rule, shape of the double bond, geometric isomerism, cis-trans isomers and E-Z nomenclature, catalytic hydrogenation, oxidative cleavage of double bond including ozonolysis.</td>
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<td></td>
<td>- Alkyne reactions treated briefly as a simple extension of alkene reactions, acidity of alkynes and nucleophilic character of the alkyne anion.</td>
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<td>- Alkyl halides, idea of leaving group, introduction to the use of curly arrows in representing mechanism, idea of nucleophiles and electrophiles, nucleophilic substitutions, SN1 and SN2 mechanisms, carbocations, dehydrohalogenation, elimination mechanisms E1 and E2 emphasising common intermediate for SN1 and E1, direction of elimination, Saytzeff rule, organo lithium and Grignard reagents as carbon nucleophiles.</td>
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<td></td>
<td>- Alcohols, hydrogen bonds, differences between primary secondary and tertiary, amphoteric nature of the OH group, alkoxides, mechanism of dehydration, oxidation.</td>
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<tr>
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<td>- Amines as bases and as nucleophiles.</td>
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<td>- Aldehydes and ketones, nucleophilic attack on the carbonyl carbon, cyanohydrins, oximes, hydrazones, Grignard products, acetals and the mechanism of their formation, oxidation and reduction of the carbonyl group, keto-enol tautomerism, the enolate anion, resonance, haloform reaction, aldol condensation.</td>
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<tr>
<td>9-12 (14 L)</td>
<td><strong>Chemistry of the Elements</strong></td>
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<td>• Carboxylic acids, acid strength, carboxylate anions, esters, acid halides, acid anhydrides, amides, emphasis on electrophilic nature of carbonyl group, mechanism of esterification and hydrolysis.</td>
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<tr>
<td></td>
<td>• Hydrogen and the s-Block Elements. Some important hydrogen compounds. Group I: the alkali metals. Group II: the alkaline earth metals. Some important Group II compounds.</td>
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<tr>
<td></td>
<td>• The d-Block Elements. Important properties of d-block elements and their compounds. Trends in physical properties. Reactions of d-block complexes.</td>
</tr>
</tbody>
</table>

**Reading list/ Indicative Resources**

- Chemistry & Chemical Reactivity Hardcover by Paul Treichel, John Kotz, John Townsend, David Treichel; Publisher: Brooks Cole; 9 ed.
- Organic Chemistry, by Jonathan Clayden and Nick Greeves; Publisher: OUP Oxford; 2 ed.

**Methods of Teaching and Student Learning**

A mixture of lectures, tutorials and hands-on laboratory practicals are used in the delivery of this module. The practical sessions are formatted in order to further clarify concepts thus reinforcing learning. A weekly hour of tutorial problem solving activities provides an additional opportunity for the lecturer to assess understanding and gauge the knowledge level of the students.

**Learning outcomes**

On completion of this module the student should be able to:

- Identify and explain bonding, hybridisation and mechanisms.
- Describe and explain the chemistry of functional groups (alkanes, alkenes and alkynes, aromatics, alkylhalides, alcohol, aldehydes, ketones and amines) and their applications.
- Analyse and discriminate between mechanisms in terms of the inherent reactivity/polarisation etc. of the two reaction components.
- Identify and classify chiral centres in organic molecules.
- Describe the chemical and physical properties of elements as a function of their position in the periodic table.
- Determine and explain the origin of the trends within groups and across periods of the properties of elements in the periodic table.
- Describe the typical structures of some common compounds of the main group elements.
• Classify elements as metallic/metalloid/non-metallic and contrast their characteristic properties.
• Explain the practical and industrial uses of key elements and compounds, and relate these to their properties.

Module Prerequisite:
CHU11101 General and Physical Chemistry (First Semester)

Assessment details:
This module will be examined via a combination of in-course assessments (25% of the final mark) and a 3 h paper at the end of semester 2 (75% of the final mark).

Course Director:
Professor Eoin Scanlan
E-mail: eoin.scanlan@tcd.ie
Phone: 01 896 2514

Coordinator Freshman Teaching
Dr Noelle Scully
E-mail: jfchem@tcd.ie
Phone: 01 896 1972

Senior Executive Officer
Ms Ann-Marie Farrell
E-mail: farrea25@tcd.ie
Phone: 01 896 1726
More than 7 billion people now inhabit the Earth and no corner of the planet is unaffected by human activity. The rise of our species has been fuelled by our ability to access planetary storehouses of energy and employ this to manipulate the environments around us. The global-scale of human impacts has led some to suggest we are entering a new era of Earth history - the Anthropocene. Dealing with the effects of environmental and climate change is one of the most significant challenges that our species faces in the 21st century.

This module provides a foundation for understanding global environmental issues by considering the Earth as an interconnected system in which matter and energy are exchanged between the Geosphere, Biosphere, Atmosphere, Hydrosphere and the Anthroposphere. It considers the life-support systems of ‘spaceship Earth’ and aims to provide a theoretical basis for evaluating the role of humans as agents of climate and environmental change.

Module learning aims
To provide foundation-level knowledge of:
- Fundamental concepts of Earth systems science and the theoretical basis of the ‘systems approach’ in Geography and Geoscience
- Character and scope of Earth’s principal sub-systems: Geosphere, Hydrosphere, Atmosphere, Biosphere and Anthroposphere
- Composition / structure of the solid Earth (Geosphere) and the principal processes / drivers responsible for its formation and evolution
- Composition / structure of atmosphere and ocean, the physical processes / drivers of their circulation, and the nature of coupling between them
- Weather and climate at a global scale including climate change past, present and future
- Biogeochemical cycling and the role of interconnected biotic and abiotic systems in the maintenance of life on Earth
- Ecological and historical biogeography including fundamentals of ecology, evolution and extinction
- Nature and scope of human impacts on the Earth system including the ‘Anthropocene’ concept

To develop the following skills & graduate attributes:
- Digital skills to manipulate and analyse geographical data, including use of Google Earth and Excel
- Self-motivated and reflective approach to independent learning, including completion of assigned reading, activities and formative assessment
- Make connections between a student’s core subject areas and the field of geography & geoscience
Module Learning Outcomes
On successful completion of this module students will be able to:

- Outline the fundamental concepts of Earth Systems Science with reference to its major subsystems: Geosphere, Biosphere, Atmosphere, Hydrosphere and Anthroposphere
- Illustrate how material and energy are cycled through the Earth system
- Describe the links between biotic and abiotic systems and their role in maintaining a habitable planet
- Apply an Earth Systems approach to describe the phenomena of environmental and climate change
- Discriminate between ‘weather’ and ‘climate’ and situate concerns about current climate change in a longer-term (geological) context
- Identify how human activities modify Earth System function
- Make links between Earth Systems Science and topics covered in their chosen field of study

Contact hours: 22 hours

Recommended Reading List:

Assessment Details: 100% continuous assessment via in-course tests and assignments.

Module Website: Blackboard
GSU11005: Introduction to Geology: A Beginners Guide to Planet Earth

Semester 2, 10 credits

From the vastness of space, to the microscopic crystal structure of minerals; from events which take billions of years, like galaxy formation, to volcanic eruptions which may last only minutes or seconds. Geology, or Earth Science, is the all-encompassing study of Planet Earth. Geology sets out to investigate the origin and development of the planet, the natural principles that govern it, the processes that act in it, on it, and around it, and finally the life that has evolved with it. Many sciences are conducted in the laboratory, but to a geologist, the Earth itself is the laboratory.

The module is organised into two main themes. Firstly, we will look at ‘Earth In Space’. We live on a dynamic and ever-changing planet, where the surface is constantly being destroyed and renewed. This theme looks at the origin of the Earth, what it’s made of and the processes at work, inside and out, which drive this change. The second theme, ‘Earth In Time’, then focuses on the evolution of the planet over time, and the life that has evolved with it. Earth has been around for just over 4,500 000 000 years, and remarkably, we have evidence that life has existed for at least 3,800 000 000 of those years. There are times in Earth’s history when geological events have changed the course of biological evolution. And, perhaps more intriguingly, there are times when life has changed the way the planet operates. So, this theme of Earth and Life evolving together through geological time is illustrated by looking at eight key episodes in Earth’s history, without which, we simply wouldn’t be here.

Module learning aims
To provide foundation-level knowledge of:

- Fundamental concepts and principal methods employed in the science of geology
- Planetary origins and evolution of planet Earth
- Rock types, composition, classification and transformation
- Structure of planet Earth and plate tectonic theory
- The distribution and causes of geohazards (volcanoes, earthquakes, tsunamis)
- Fossil evidence of the origins and evolution of life on Earth
- Geological evidence of past environments and climatic conditions
- Economic geology and the nature of geological resources

To develop the following skills & graduate attributes

- Written and digital / analytical skills
- Critical thinking
- Make connections between a student’s core subject areas and the science of Geology

Module learning outcomes
On successful completion of this module students will be able to:

- Outline the origin and evolution of planet Earth
- Describe and illustrate the dynamic nature of planet Earth with reference to specific geological processes
- Describe the origins of life on Earth and list the major evolutionary episodes evident in the fossil record
- Explain the links between the evolution of life and environmental conditions on planet Earth
- Outline the geological history of the island of Ireland
- Make basic geological observations, measurements and interpretations in the field and laboratory
Recommended Reading List:

Assessment details: 50% examination; 50% continuous assessment via in-course tests and assignments.

Module website: https://www.tcd.ie/Geology/undergraduate/modules/year1/

Course Director TR062 Geography and Geoscience
Professor Robin Edwards E-mail: Robin.Edwards@tcd.ie
Phone: 01 896 1713

Executive Officers:
Geography Department E-mail: geog@tcd.ie
Ph: 01 896 1576

Sarah Guerin E-mail: earth@tcd.ie
Geology Department Ph: 01 896 1074
Dates to Note:

Freshers Orientation Week: 28th September 2020 – 2nd October 2020

**Academic Year Structure 2020/21**

**Key Dates:**

- **Freshers/Orientation Week:** Monday 28th September to Friday 2nd October 2020
- **Semester One 2020 (Michaelmas Term 2020):** Monday 5th October 2020 to Friday 18th December 2020
- **Trinity week:** Monday 26th April to Friday 30th April 2021
- **Semester 2 (Hilary Term):** Monday 1st February 2021 – Friday 23rd April 2021
- **Study/Review Week:** Monday 15th March to Friday 19th March 2021
- **Formal Assessment weeks**
  - *Semester 1 examinations:* Monday 11th January to Friday 22nd January 2021
  - *Semester 2 examinations:* Monday 10th May to Friday 21st May 2021
- *Extra contingency days may be required outside of the formal assessment / reassessment weeks.*
Closing Dates for Course Transfer

If you decide to transfer out of your course altogether, you must submit an application for transfer of course to the Academic Registry, following discussion with your tutor. Decisions are based on a) the availability of places, and b) the entry qualifications of the transfer applicant. It may not be possible to permit transfers to subjects which already have a full complement of students. Further details are available on the following link:

http://www.tcd.ie/study/apply/making-an-application/undergraduate/index.php

Students may not register or attend a course until their application to transfer has been formally approved by the Senior Lecturer.

Progression and Awards

Information on progression and awards can be found via the following webpage:

https://www.tcd.ie/TEP/assets/Docs/factsheet_students_progression_awards.pdf

Attendance

All students should enter into residence in or near Dublin and must begin attendance at the College not later than the first day of teaching term, and may not go out of residence before the last day of teaching term, unless they have previously obtained permission from the Senior Lecturer through their tutor.

Students must attend College during the teaching term. They must take part fully in the academic work of their class throughout the period of their course. Lecture timetables are published through my.tcd.ie and on school or department notice-boards before the beginning of Michaelmas teaching term. The onus lies on students to inform themselves of the dates, times and venues of their lectures and other forms of teaching by consulting these timetables.

The requirements for attendance at lectures and tutorials vary between the different faculties, schools and departments. Attendance is compulsory for Junior Freshman in all subjects. The school, department or course office, whichever is relevant, publishes its requirements for attendance at lectures and tutorials on notice-boards, and/or in handbooks and elsewhere, as appropriate.
Absence from College – Medical and Absence Certificates

Medical Certificates
Where a student misses an assigned laboratory practical class through illness, they should (a) submit a Medical Certificate to the Science Course office on the day of their return to College and (b) inform the laboratory practical supervisor of their absence at the next session.

- Science Medical Certificate Form (use with med cert from doctor) – Available from Science Course Office

For periods of illness of three days or less (but no more than seven days in any year) a student may ‘self-certify’ their illness on the forms supplied, again to the Science Course Office on the day of their return to College.

- Science Medical Self Certification Form (use for 3 days med not covered by doctor) – Available from the Science Course Office

Other absences
Students who require to be absent from a laboratory practical classes (with or without an associated assessment) for any other reason, such as a sporting event, should inform the appropriate module coordinator well in advance of the event (preferably a week beforehand).

- Science Absence from College Form, Sport or Other – Available from the Science Course Office

Where possible they will be assigned to an alternative laboratory practical session, but if that is not possible, and the justification for the absence is considered legitimate, they may be treated in the same manner as students submitting medical certificates (i.e. assigned an alternative assessment for one missed or awarded a pro-rata/pass mark). This is decided by the individual Disciplines concerned (i.e. Biology, Chemistry, Physics etc.) not the Science Course Office.

Excuses for absence, presented after the event, will not be entertained. Students who anticipate that their sporting commitments may necessitate more than the occasional absence from College (e.g. Sport Scholars, etc.) should discuss their situation with their tutor and the Associate Dean of Undergraduate Science Education (ADUSE).

Non-satisfactory attendance and course work
All students must fulfil the requirements of the school or department, as appropriate, with regard to attendance and course work. Where specific requirements are not stated, students may be deemed non-satisfactory if they miss more than a third of their course of study or fail to submit a third of the required course work in any term. Further information on non-satisfactory attendance and course work may be found via the following webpage:

https://www.tcd.ie/undergraduate-studies/academic-progress/attendance-course-work.php
Plagiarism- 2020/21

It is important to emphasise that all students, i.e., undergraduate, postgraduate, new entrants and existing students, will be required to complete the online tutorial ‘Ready, Steady, Write’. Students must ensure that the cover sheets they complete when submitting assessed work, contain the following declaration:

I have read and I understand the plagiarism provisions in the General Regulations of the University Calendar for the current year, found at:  http://www.tcd.ie/calendar

I have also completed the Online Tutorial on avoiding plagiarism ‘Ready, Steady, Write’, located at http://tcd-ie.libguides.com/plagiarism/ready-steady-write

Students should read the items listed below to ensure that they understand plagiarism.

2. The 2020-21 Calendar entry on plagiarism; Plagiarism
3. Guidelines on the appropriate methodology for the kind of work that students will be expected to undertake. Providing discipline specific examples of good academic practice for referencing is very helpful for students. We would like to draw your attention to the 2020-21 Calendar entry on plagiarism which states that “all Schools and departments must include in their handbooks or other literature given to students, guidelines on the appropriate methodology for the kind of work that students will be expected to undertake”;
4. A statement informing all students that they must complete the online tutorial on avoiding plagiarism ‘Ready, Steady, Write’, located at http://tcd-ie.libguides.com/plagiarism/ready-steady-write
5. The template of the coversheet/s which students must complete and attach to work submitted in hard or soft copy or via Blackboard. NB. The coversheet must include the declaration noted above.

**Trinity Tutorial Service**

The Tutorial Service is unique, confidential and available to all undergraduate students offering student support in all aspects of College life. The Tutorial Service is supported and co-ordinated by the Senior Tutor’s Office which is located on the ground floor in House 27.

**Opening Hours**
The Senior Tutors Office is open Monday - Friday from 9am - 5.30pm. Closed for lunch from 1-2pm.

**Appointments**
If you require specific advice or would like a confidential meeting with the Senior Tutor, you can make an appointment by telephoning +353 1 896 2551 or by emailing stosec@tcd.ie

**What is a Tutor?**
A Tutor is a member of the academic staff who is appointed to look after the general welfare and development of the students in his/her care. Whilst the Tutor may be one of your lecturers, this is not always the case as the role of the College Tutor is quite separate from the teaching role.

**When should I go to see my Tutor?**
Whenever you are worried or concerned about any aspect of College life or your personal life, in particular if it is affecting your academic work. Everything you say to your Tutor is in strict confidence. Unless you give him/her permission to do so, s/he will not give any information to anybody else, whether inside College or outside (to your parents/family for example). Your Tutor can only help you if s/he knows you are facing difficulties, so if you are worried about anything go and see your Tutor before things get out of hand.

Further information on the Senior Tutors Office and College Tutors may be found via the following webpage: **Senior Tutor’s Office - https://www.tcd.ie/seniortutor/students/undergraduate/**

**Disability Services**

The Disability Service aims to provide appropriate advice, support and information to help students and staff with disabilities. The Disability Service has in place a range of supports to ensure that students with disabilities have full access to the same facilities for study and recreation as their peers. Most students registering with the Disability Service request access to a range of supports that help the student reach their full potential while studying. Most students’ needs are accommodated through these supports. The student decides what level of support they require.

Further information on the support available may be found via the following link: **https://www.tcd.ie/disability/services/**

For contact information or to make an appointment please contact the Disability Services – contact details are available via the following webpage: **https://www.tcd.ie/disability/contact/**
Student Counselling

The Student Counselling Service is here to help you to manage any difficulties you are experiencing so you can enjoy and fully participate in your time here at College.

If you wish to make an appointment with the Student Counselling Service, please consider one of the options below. If you have any other queries you can call into reception on the 3rd floor of 7-9 South Leinster Street or contact us on:

Phone: (01) 8961407
Email: student-counselling@tcd.ie

For further information visit the following webpage:
https://www.tcd.ie/Student_Counselling/

Useful College Websites:

Orientation – Student Life
https://www.tcd.ie/students/orientation/

In the first few weeks at College, you will hear an array of abbreviations, titles and place names. So visit the jargon buster page: https://www.tcd.ie/students/jargon-buster/

Student life offers information on Supports and Services, Clubs and Societies, Student Unions etc., https://www.tcd.ie/students/

For information on Registration, Fees, Grants, ID Cards etc. visit the Academic Registry (AR) in the Watts Building or visit the AR website: https://www.tcd.ie/academicregistry/
TR063: Physical Sciences
Contacts:

Physical Sciences Course Director
Professor Cormac McGuinness
Ph: 01 896 3547
E-mail: Cormac.McGuinness@tcd.ie

Junior Freshman Physics Coordinator
Professor Matthias Moebius
Ph: 01 896 1055
E-mail: mobiusm@tcd.ie

School of Physics Executive Officer
Ms Helen O’Halloran
Ph: 01 896 4141
E-mail: physics@tcd.ie

Science Course Office
Professor Áine Kelly
Associate Dean of Undergraduate Science Education
Ph: 01 896 2025
E-mail: aduse@tcd.ie

Ms Anne O’Reilly
Science Course Administrator
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E-mail: science@tcd.ie

Ms Ann Marie Brady
Senior Executive Officer
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E-mail: ennisa@tcd.ie

Ms Agnes Gogan
Senior Executive Officer
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Ms Lucy Martin
Executive Officer
Ph: 01 896 2022
Email: martinl3@tcd.ie

Ms. Mary Pat O’Sullivan
Executive Officer
E-mail: mpsullvn@tcd.ie
Ph: 01 8961970

Ms Eva Page
Global Officer: Life and Biosciences
E-mail: eva.page@tcd.ie
Ph: 01 896 2799
### Teaching Term Dates 2020-2021

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<td><strong>Monday 1st February 2021 - Friday 23rd April 2021</strong></td>
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<tr>
<td>Teaching wk 1</td>
<td>Teaching wk 1</td>
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<tr>
<td>Week 05</td>
<td>Week 23</td>
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<tr>
<td>28th Sept – 2nd Oct</td>
<td>1st Feb – 5th Feb</td>
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<tr>
<td>Teaching wk 2</td>
<td>Teaching wk 2</td>
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<tr>
<td>Week 06</td>
<td>Week 24</td>
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<td>5th Oct – Oct 9th</td>
<td>8th Feb – 12th Feb</td>
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<tr>
<td>Teaching wk 3</td>
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<td>Week 07</td>
<td>Week 25</td>
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<td>12th Oct – 17th Oct</td>
<td>15th Feb – 19th Feb</td>
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<tr>
<td>Teaching wk 4</td>
<td>Teaching wk 4</td>
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<tr>
<td>Week 08</td>
<td>Week 26</td>
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<td>19th Oct – 23rd Oct</td>
<td>22nd Feb – 26th Feb</td>
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<td>Teaching wk 5</td>
<td>Teaching wk 5</td>
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<tr>
<td>Week 09</td>
<td>Week 27</td>
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<td>*26th Oct – 30th Oct</td>
<td>1st Mar – 5th Mar</td>
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<tr>
<td>Teaching wk 6</td>
<td>Teaching wk 6</td>
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<td>Week 10</td>
<td>Week 28</td>
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<tr>
<td>2nd Nov – 6th Nov</td>
<td>8th Mar – 12th Mar</td>
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<tr>
<td><strong>Study week</strong></td>
<td><strong>Study week</strong></td>
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<tr>
<td><strong>Week 11</strong></td>
<td><strong>Week 29</strong></td>
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<tr>
<td>9th Nov – 13th Nov</td>
<td><strong>15th Mar – 19th Mar</strong></td>
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<tr>
<td>Teaching wk 8</td>
<td>Teaching wk 8</td>
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<td>Week 12</td>
<td>Week 30</td>
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<tr>
<td>16th Nov – 20th Nov</td>
<td>22nd Mar – 26th Mar</td>
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<tr>
<td>Teaching wk 9</td>
<td>Teaching wk 9</td>
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<tr>
<td>Week 13</td>
<td>Week 31</td>
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<tr>
<td>23rd Nov – 27th Nov</td>
<td>29th Mar – 2nd Apr</td>
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<td>Teaching wk 10</td>
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<td>Week 14</td>
<td>Week 32</td>
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<td>30th Nov – 4th Dec</td>
<td><strong>5th Apr – 9th Apr</strong></td>
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<td>Teaching wk 11</td>
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<td>Week 15</td>
<td>Week 33</td>
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<td>7th Dec – 11th Dec</td>
<td>12th Apr – 15th Apr</td>
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<tr>
<td>Teaching wk 12</td>
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<td>Week 16</td>
<td>Week 34</td>
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<td>14th Dec – 18th Dec</td>
<td>19th Apr – 23rd Apr</td>
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* Monday 26th October 2020 Bank Holiday - College closed

** Wednesday 17th March 2021 St Patricks Day - College closed

***Monday 5th April 2021 - Easter Monday - College Closed
Graduate Attributes

The Trinity Graduate Attributes represent the qualities, skills and behaviours that you will have the opportunity to develop as a Trinity student over your entire university experience, in other words, not only in the classroom, but also through engagement in co- and extra-curricular activities (such as summer work placements, internships, or volunteering).

The four Trinity Graduate Attributes are:

- To Think Independently
- To Act Responsibly
- To Develop Continuously
- To Communicate Effectively

Why are the Graduate Attributes important?

The Trinity Graduate Attributes will enhance your personal, professional and intellectual development. They will also help to prepare you for lifelong learning and for the challenges of living and working in an increasingly complex and changing world.

The Graduate Attributes will enhance your employability. Whilst your degree remains fundamental, also being able to demonstrate these Graduate Attributes will help you to differentiate yourself as they encapsulate the kinds of transversal skills and abilities, which employers are looking for.

How will I develop these Graduate Attributes?

Many of the Graduate Attributes are ‘slow learned’, in other words, you will develop them over the four or five years of your programme of study.

They are embedded in the curriculum and in assessments, for example, through undertaking independent research for your final year project, giving presentations and engaging in group work.

You will also develop them through the co-curricular and extra-curricular activities. If you help to run a club or society you will be improving your leadership skills, or if you play a sport you are building your communication and team-work skills.
# Appendix 1

## Appendix 1: General Information

<table>
<thead>
<tr>
<th>ITEM</th>
<th>REFERENCE/Source</th>
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<tbody>
<tr>
<td><strong>General Regulations</strong></td>
<td>Calendar, Part II - General Regulations and Information, Section II, Item 12: <a href="https://www.tcd.ie/calendar/undergraduate-studies/general-regulations-and-information.pdf">https://www.tcd.ie/calendar/undergraduate-studies/general-regulations-and-information.pdf</a></td>
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<td>Absence from Examinations Calendar, Part II, General Regulations and Information, Section II, Item 35 Calendar, Part III, Section 3.5</td>
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<td><strong>General Information</strong></td>
<td>Timetable are available via my.tcd.ie portal: <a href="https://my.tcd.ie/urd/sits.urd/run/siw_lgn">https://my.tcd.ie/urd/sits.urd/run/siw_lgn</a></td>
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<td>Science Foundation Scholarship information sheet:</td>
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## Appendix 1: General Information

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<th>Teaching and Learning</th>
<th>Academic Policies: <a href="https://www.tcd.ie/teaching-learning/academic-policies/">https://www.tcd.ie/teaching-learning/academic-policies/</a></th>
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<td>Student Complaints Procedure: <a href="https://www.tcd.ie/about/policies/160722_Student%20Complaints%20Procedure_PUB.pdf">https://www.tcd.ie/about/policies/160722_Student%20Complaints%20Procedure_PUB.pdf</a></td>
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<td>Student Services Booklet: <a href="http://www.tcd.ie/students/assets/pdf/Student%20Services%20Booklet%20(web%20version).pdf">www.tcd.ie/students/assets/pdf/Student%20Services%20Booklet%20(web%20version).pdf</a></td>
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<td>Senior Tutor &amp; Tutorial Service <a href="http://www.tcd.ie/students/assets/pdf/Student%20Services%20Booklet%20(web%20version).pdf">www.tcd.ie/students/assets/pdf/Student%20Services%20Booklet%20(web%20version).pdf</a></td>
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<td>Mature Student Office <a href="https://www.tcd.ie/maturestudents/">https://www.tcd.ie/maturestudents/</a></td>
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<th>Co-curricular activities</th>
<th>Central Societies Committee: <a href="https://www.tcd.ie/calendar/general-information/students-unions-societies-and-clubs.pdf">https://www.tcd.ie/calendar/general-information/students-unions-societies-and-clubs.pdf</a></th>
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<td>DUCAC: <a href="https://www.tcd.ie/Sport/student-sport/ducac/?nodeId=94&amp;title=Sports_Clubs">https://www.tcd.ie/Sport/student-sport/ducac/?nodeId=94&amp;title=Sports_Clubs</a></td>
</tr>
</tbody>
</table>
## Appendix 1: General Information

| Information on TCDSU and GSU, Including student representative structures | TCDSU  
https://www.tcdsu.org/  
TCDSU Student Representation Overview  
https://www.tcdsu.org/aboutus  
TCD GSU  
https://www.tcdgsu.ie/  
GSU - Student Representation Overview  
https://www.tcdgsu.ie/becomearep/ |
|---|---|
| Emergency Procedure | In the event of an emergency, **dial Security Services on extension 1999**  
Security Services provide a 24-hour service to the college community, 365 days a year. They are the liaison to the Fire, Garda and Ambulance services and all staff and students are advised to always telephone extension 1999 (+353 1 896 1999) in case of an emergency.  
Should you require any emergency or rescue services on campus, you must contact Security Services. This includes chemical spills, personal injury or first aid assistance.  
It is recommended that all students save at least one emergency contact in their phone under ICE (In Case of Emergency). |

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**NOTE:** All of the information contained in this booklet is accurate at time of publication. However, the Science Course Office reserves the right to modify information, dates and times as necessary. Students will be notified of any changes via e-mail and the Science webpage.
Science Course Office

Faculty of Engineering, Mathematics and Science Trinity College Dublin 2, Ireland.

Oifig na gCúrsaí Êolaíochta Dámh na hinne-altóireachta, na Matamaitce agus na hÉolaíochta Ollscoil Átha Cliath, Coláiste na Tríonóide Baile Átha Cliath 2. Eire.

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E-mail: science@tcd.ie
Web: www.tcd.ie/Science
tcd.ie/science